



Use of the Design of Experiments to Develop a Scalable Route to a Key Chirally Pure Arylpiperazine

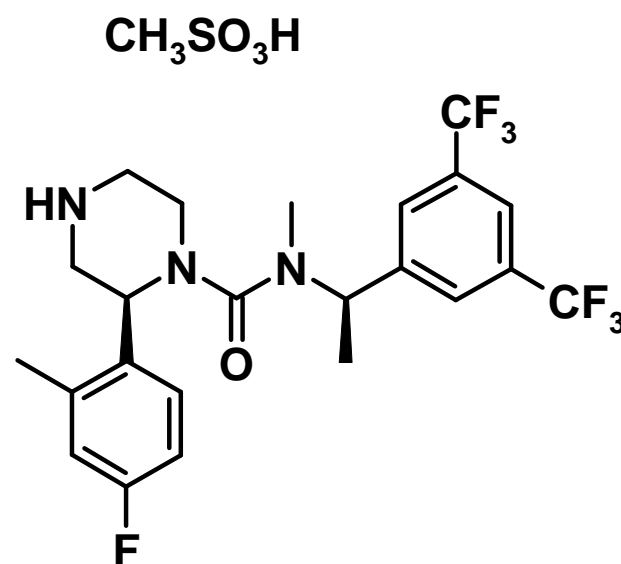
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Introduction

- **A Design of Experiments applied to a chemical process is presented**
- **The synthesis of a chemical intermediate was studied in lab**
- **Some issues were encountered when the process was scaled up in the Pilot Plant**
- **DoE and computer modeling were used to understand the process and define new conditions**
- **New conditions were successfully applied to the following Pilot Plant campaign**

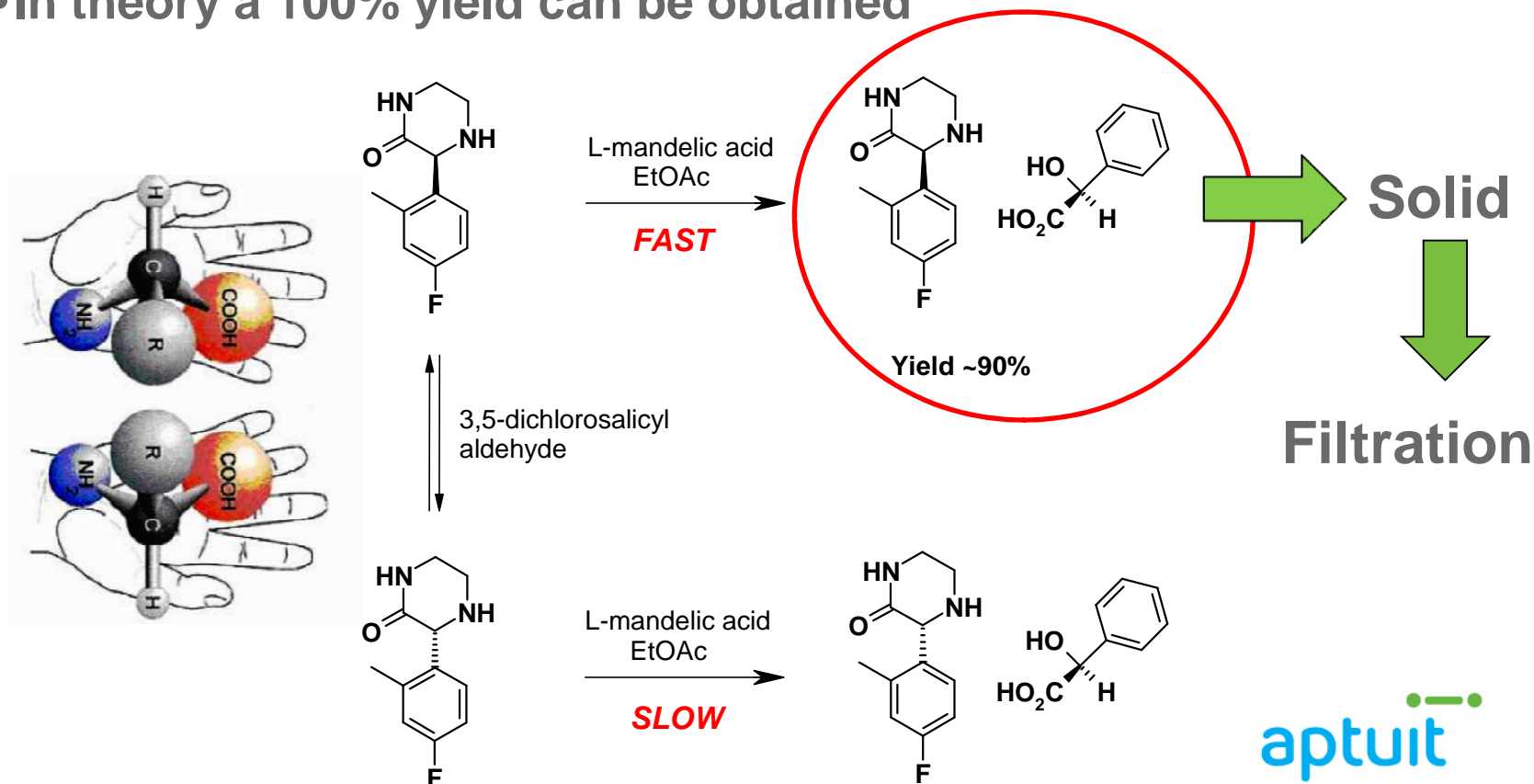
Vestipitant Mesylate

- NK-1 antagonist
- Total synthesis: 3 Stages, 9 Steps
- 2 chiral centers: 4 possible isomers
- Final product isolated as mesylate salt
- Dynamic Kinetic Resolution (DKR) is a key step in the synthesis of Vestipitant Mesylate



Dynamic Kinetic Resolution (DKR)

- Well known industrial method
- It combines the resolution step with an *in situ* racemisation
- In theory a 100% yield can be obtained



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Lab Optimization

- The reaction was studied in laboratory using “conventional” methods (no statistical tools)
- Very viscous suspension at the beginning, more fluid after stirring
- Scaled on a 5 L scale
- No main issues encountered
- Yield= ca. 90% of the pure enantiomer



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Pilot Plant Issues

- Very difficult azeotropic removal of the water to the target level
- Interaction of water with the reaction
- Slurry difficult to stir
- A considerable amount of product adhering to the reactor walls
- Yield decreased from 90% to *ca. 50%!!!*

Chemical problem or Mechanical problem?



Back to the lab

- Further understanding and optimization needed
- Brainstorming with different functions (chemists, engineers, analysts etc.)
- DoE required
- Parallel equipment used:
 - Able to mimic pilot plant conditions
 - Mechanical overhead stirring
 - Four reactions at the same time
- An initial screening with many parameters
- Identified three parameters as possible causes
 - Aldehyde amount
 - Water amount
 - Stirring rate



Design of Experiment

- Three parameters considered
- Three responses:
 - Solid recovery (yield)
 - Solid purity (enantiomeric excess)
 - Total recovery (solid+solution)

		Number of Factors											
		2	3	4	5	6	7	8	9	10	11	12	13
Experiments	4	Full	1/2 Fract.										
	8		Full	1/2 Fract.	1/4 Fract.	1/8 Fract.	1/16 Fract.						
	16			Full	1/2 Fract.	1/4 Fract.	1/8 Fract.	1/16 Fract.	1/32 Fract.	1/64 Fract.	1/128 Fract.	1/256 Fract.	1/512 Fract.
	32				Full	1/2 Fract.	1/4 Fract.	1/8 Fract.	1/16 Fract.	1/32 Fract.	1/64 Fract.	1/128 Fract.	1/256 Fract.
	64					Full	1/2 Fract.	1/4 Fract.	1/8 Fract.	1/16 Fract.	1/32 Fract.	1/64 Fract.	1/128 Fract.
	128						Full	1/2 Fract.	1/4 Fract.	1/8 Fract.	1/16 Fract.	1/32 Fract.	1/64 Fract.
	256							Full	1/2 Fract.	1/4 Fract.	1/8 Fract.	1/16 Fract.	1/32 Fract.

- Fractional Factorial Design (Full) with three Blocks

- Block 2:

- Two center points, to evaluate the curvature
- Two repeated points (factorial), to estimate the block effect and the pure error

- Twelve reactions

Result output 1: Solid Recovery

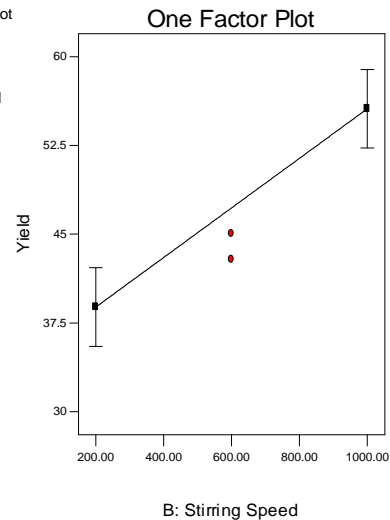
DESIGN-EXPERT Plot

Yield

X = B: Stirring Speed

• Design Points

Actual Factors
A: Aldehyde = 0.07
C: Water = 0.27



Design-Expert® Software

Factor Coding: Actual

Yield

• Design points above predicted value

○ Design points below predicted value

58

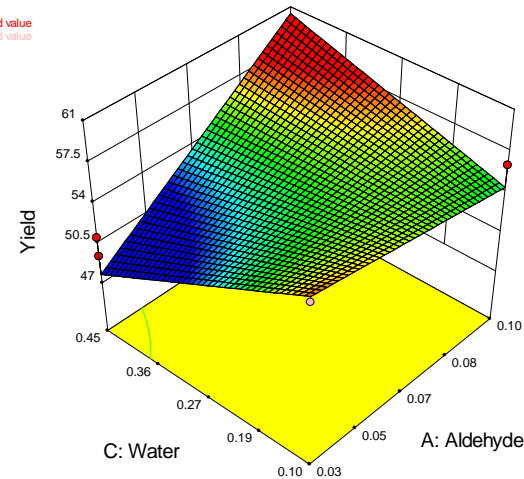
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X1 = A: Aldehyde

X2 = C: Water

Actual Factor

B: Stirring Speed = 1000.00



B) Stirring rate:

- the most significant factor
- high yields were obtained with high stirring speeds

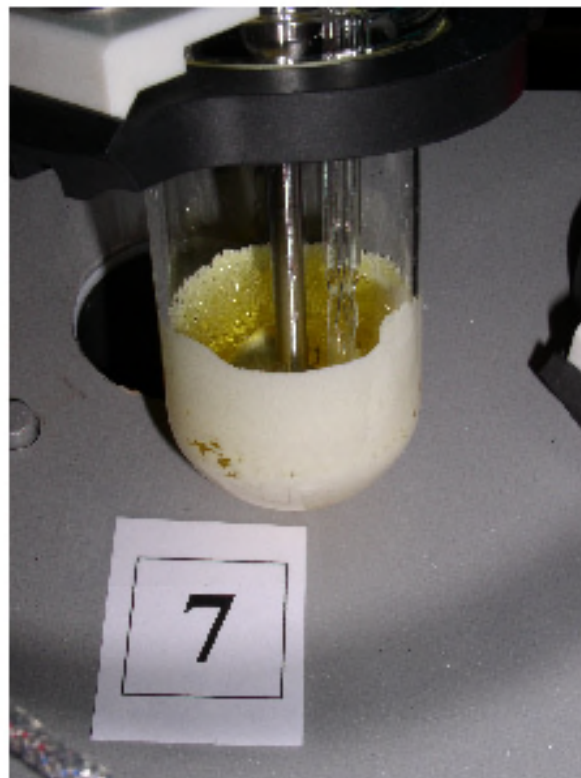
•Curvature not significant

•Mechanical and Chemical effects

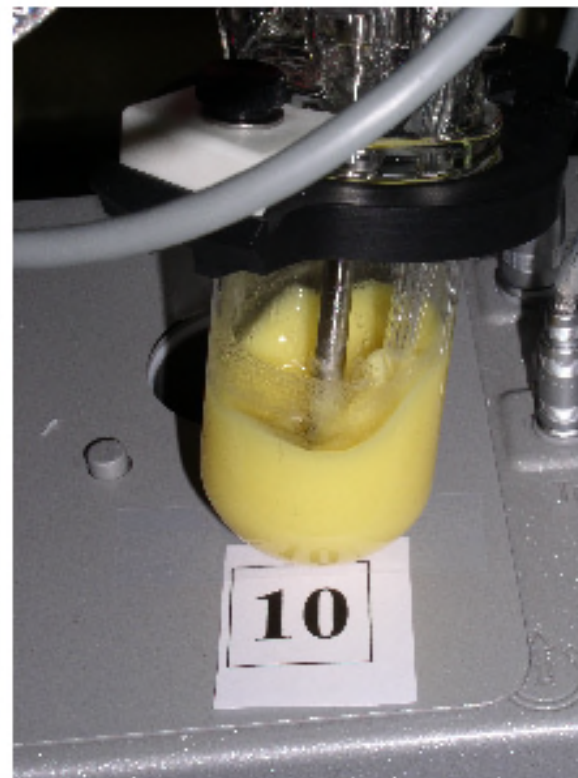
AC) Interaction Aldehyde/Water

- a larger amount of water was buffered by a larger amount of aldehyde

Effect of Aldehyde on the solid



Run7
Aldehyde=0.033wt
Stirring=250rpm
Yield=0%



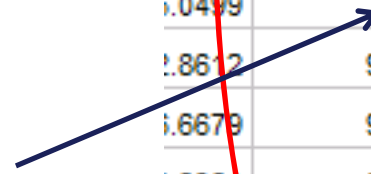
Run10
Aldehyde=0.1wt
Stirring=1000rpm
Yield=56%

Result output 2: Solid purity

- Required $\geq 99\%$
- Response range between 99.0 and 99.75%
- Considered robust!

Response 1 Std	Response 2 e.e.	Re R
1.4496	99.51	
1.4375	99.24	
1.4553	99.37	
1.9699	99.49	
1.7322	99.4	
1.1387	99.27	
1.0499	99	
1.8612	99.39	
1.6679	99.13	
1.8361	99.47	
1.6173	99.75	
1.7023	99.31	

Lowest value



Result output 3: Total Recovery (Solid+Solution)

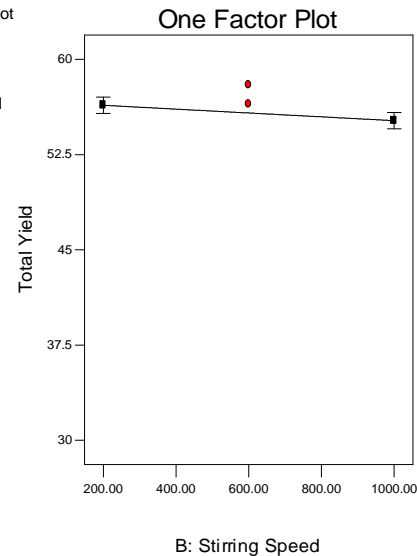
DESIGN-EXPERT Plot

Total Yield

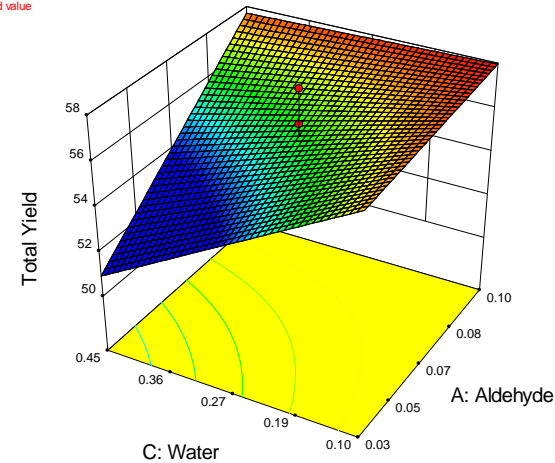
X = B: Stirring Speed

• Design Points

Actual Factors
A: Aldehyde = 0.07
C: Water = 0.27



Design-Expert® Software
Factor Coding: Actual
Total Yield
• Design points above predicted value
58
54
X1 = A: Aldehyde
X2 = C: Water
Actual Factor
B: Stirring Speed = 600.00



B) stirring rate:

- Not significant

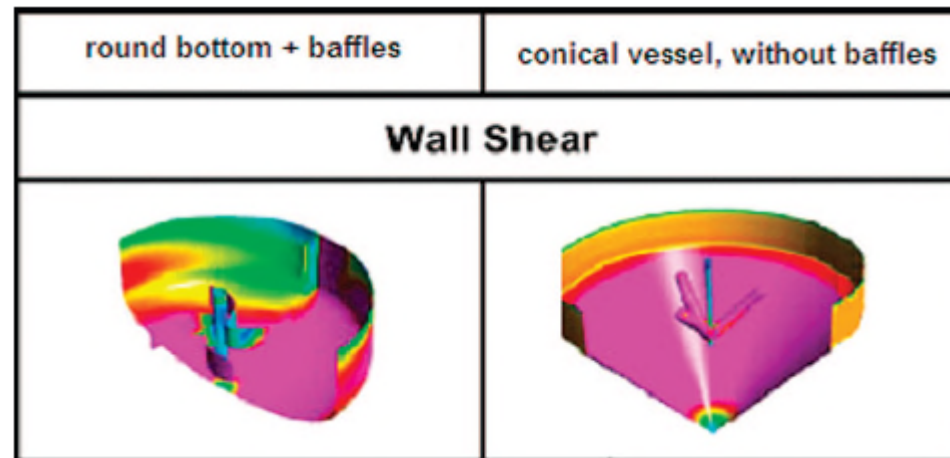
- Curvature not significant
- Only Chemical effects

AC) interaction Aldehyde/Water

- The negative effect of higher water amount was balanced by a higher quantity of aldehyde

Computer Modeling

- Design an appropriate reactor configuration
- Ensure efficient mixing.



- Round bottomed vessel with baffles (initially employed)
 - Dead mixing zones
 - Encrustation
- Conical vessel without baffles
 - No dead mixing zone
 - Ideal for thick suspensions.

Test on the Pilot Plant

- The new process was repeated in the pilot plant using:
 - Suggested vessel configuration
 - More aldehyde
 - Proper stirring rate
- Yield and Purity reproduced



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Conclusions

- **Pilot Plant issues**



- **Brainstorming to decide how to approach the problem**



- **DoE to understand the problem encountered in the Pilot Plant**
- **Parallel equipment used to mimic on a small scale the problems**



- **Identified parameters affecting the final yield**
- **Computer modeling to identify proper vessel configuration**



- **New condition successfully applied on the Pilot Plant**

Acknowledgements

Giuseppe Guercio: Project Leader

Sergio Bacchi, Antonella Carangio, Michael Goodyear: Chemists

Francois Richard: Process Engineer

Paolo Repeto, Mohammad Yahyah: statisticians

Giulio Camurri, Luca Martini: Analysts

Stefano Curti: Process Safety

Guercio, G.; Bacchi, S.; Goodyear, M.; Carangio, A.; Tinazzi, F. and Curti, S.

Synthesis of the NK1 Receptor Antagonist GW597599. Part 1: Development of a Scalable Route to a Key Chirally Pure Arylpiperazine

Organic Process Research & Development, vol.12, n. 6 (2008), p.1188-1194